# Optimization of TiO<sub>2</sub> Based Henna Dye Sensitized Solar Cell using Grey-Taguchi Technique

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#### Abstract

Low efficiency is a major drawback associated with Dye Sensitized Solar Cells (DSSCs). However, the efficiency can be enhanced by multi-objective optimization based on process parameters. In this work, four process variables were identified as important parameters that influence DSSC responses (efficiency, shunt and series resistance). These are thickness of the working electrode (A), annealing temperature (B), annealing time (C) and dye loading time (D). An optimal combination of the response parameters of the resulting DSSC was obtained by simultaneously adjusting the input parameters. With three different levels for each process variable, FTO glasses as components and henna leave extract as sensitizer, 9 unique DSSCs were fabricated following Taguchi Design of experiment. Each cell was illuminated with constant light intensity of 100 mW/cm<sup>2</sup> to measure the photovoltaic parameters. The optimal combination of the process parameters was determined by applying Grey Relational Analysis (GRA). The experimental best condition ( $A_3B_1C_3D_2$ ) obtained from Taguchi array corresponded to average thickness of 9.67 µm for TiO<sub>2</sub> film, annealing temperature of 300 °C, annealing time of 6 hours and dye loading time of 4 hours. With optimization, the annealing time is reduced by ~67% while other responses remain the same. The global best DSSC has efficiency, shunt and series resistance of0.66%, 24844.44  $\Omega$  and 8.4281  $\Omega$ , respectively. These indicate that the efficiency, shunt and series resistance were improved by 1.3% 12.4% and 10.8%, respectively. These results confirm that solar cell responses can be optimized significantly.

Keywords: Optimization, DSSC, Henna, efficiency, Grey-Taguchi Technique

#### 1. Introduction

Dye sensitized solar cell (DSSC), an example of organic solar cell, belongs to the 3rd generation [1]. Herein, a wide band-gap metal oxide semiconductor is sensitized by adsorbed

molecular dye within the region of visible light. Early results of electricity generation from ZnO thin film sensitized with Chlorophyllswas reported [2]. While previous results were not encouraging, Gratzel and O'Regan [4] conducted a fundamental study which showed that DSSCs could be a feasible alternative energy source. On this note, Rutheniumbased dye DSSC was developed with an efficiency of ~11.5% [5,6] while Zn-based dyes and Co-based electrolyte pair had~12% conversion efficiency [7].The efficiency of DSSC is affected by factors such as working electrode thickness, post deposition treatment of the anode and amount of dye molecules absorbed by the layer.

In most DSSCs, workingelectrodes consist of mesoporous oxide film, typically 10 µm thick, having porosity of 50% which is made up of arrays of tiny crystals measuring a few nanometers deposited on conductive glass. Oxides such as TiO<sub>2</sub>, ZnO, SnO<sub>2</sub>, NbOor chalcogenides such as cadmium selenide are the preferred photo-electrodes [8].Optimum thickness is criticalto the performance of DSSCs. With increased thickness, more dye molecules present in the electrode layer absorb sunlight, hence an increase in However, more photo-generated current generation. electrons are recombined due to longer path, thus current decreases after an optimum thickness [9]. Also, thickness dependence is a function of particle size and surface structure. For example, Ito et al [10] reported that optimal thickness for 20 nm particles is one-half of that obtained for the 42 nm particles. The internal surface areas of the nanoparticles decrease simultaneously with higher annealing temperature which affects the dye molecule adsorption as well as the light harvesting [11]. Among other methods of depositing photoconductive layer, Doctor Bladingmethod has been widely employed because of its simplicity and low cost. The thickness is controlled by the thickness of the tape guide layers.

Annealing temperature profile influencesparticle size, film porosity and consequently cell performance. A good example is the stepwise annealing of titaniawithin the temperature range of 200 - 400°C [12] and annealing temperature of up to 700°C [27-28]. Thermal annealing effects on the microstructure and dynamics of electron transport and recombination in TiO<sub>2</sub> and ZnO based DSSCs have also been investigated [13]. It was reported that electron diffusion coefficient and life time are enhanced at higher annealing temperature, resulting in more efficient electron transport within the photoelectrode [12,13]. This is attributed to highly porous and smaller grain resulted from stepped annealing. Effectively, lower Vocand higher Jscwith lower series resistance are obtained [9]. Lower Voc is attributed to more charge recombination as the electrolyte goes into the semiconductor film while higher J<sub>sc</sub> is due to more dye

absorption on the titania surface [9]. Un-stepped annealing however produced compact and largergrain.

Due to the significant role of dye, development and improvement of new families of organic dyes and metal complexes are of considerable interest. So far, the most efficient dyes are found to be Ru(II) [14, 15] and Os(II) [16]. These complexes have a number of interesting features such as good absorption, long excited lifetime and highly efficient metal-to-ligand charge transfer. They are however very and preparation technique is sophisticated. costly, Consequently, natural dyes are viable alternatives. This is because they are more available, environmental friendly and cost effective. Natural dyes extracted from fruits, leaves, roots and flowers have been proven to be efficient as photosensitizers in DSSCs [17, 18]. The amount of dye absorbed by photoconductive layer depends on thesensitization duration during fabrication. However, thisdurationis determined by the dye type. In most natural dyes, for example, anthocyaninscan be saturably absorbed by TiO<sub>2</sub> layer within minutes while dyes from other sources may take hours before fully absorbed by layer of the same thickness.

An efficient DSSC with high shunt resistance and low series resistance is desirable for greater efficiency. To achieve the best combination of these characteristics, the process variables can be optimized to deliver a simultaneous optimal response of all the desired DSSC responses using a combination of Taguchi Design of Experiment (DOE) and Grey relational analysis (GRA). This combination is one of the most widely used multi-response optimization schemes [19-20]. DOE had been widely employed to improve quality of products and lower cost of production in many fields. In this case, the influencing factors are kept at certain ranges and levels within the limit of available tools or based on literature. The levels can follow a differential or multiple increments, and not necessarily at uniform interval [19-21]. Experiments are then designed by selecting from Taguchi orthogonal array table in which the number of experiments is determined by the number of factors considered and the number of levels at which each factor is kept. The results obtained are then analysedfollowing a well-established procedure [20, 21]. The concept of GRA makes it easy to evaluate the similarity or difference between two sequences, which is helpful in solving many complex and multivariate problems [21].

In this study, we utilized the combined Taguchi approach and GRA to optimize the efficiency, shunt resistance and series

resistance of henna dye based DSSC with simultaneous consideration of semiconducting film thickness, annealing temperature, annealing time and dye loading time.

## 2. Materials and Method

In this section, we discuss the procedure for dye preparation, Taguchi experimental design technique, grey relational analysis procedure, preparation and characterization of the solar cells among others.

# 2.1. Dye Preparation

Fresh LawsoniaInermisleaves, being a common type of leaves, were collected from Botanical Garden of LadokeAkintola University of Technology Ogbomoso, Nigeria. They were washed with distilled water, airdried and ground into powder. About 10 g of the powder was poured into a bottle, and100ml of concentrated ethanol was added and stirred with a glass rod until a uniform suspension was obtained. The solution was left for 24 hours in the dark to ensure complete extraction. It was then filtered to remove solid fragments. Similarly, dyes were extracted from equal mass each of processed powdered henna leave, industrial red henna and industrial black henna. To ensure stability before use, the filtrates were kept in separate clean bottles, and were protected from direct sunlight. The extracts were optically characterized using Genway UV-VIS spectrophotometer.

# 2.2. Experimental Design

Four factors at three levels were considered for the experiment:guide layer thickness, annealing temperature, annealing time and dye loading time (Table 2). These factors and levels were chosen based on similar studies and the limit of the available experimental tools. It should be noted that these factors are quite sufficient for the current optimization process. The experimental conditions were defined by the L9 orthogonal array (Table 3).ay

 Table 2: Factor Specification

Table 3: L9 Orthogonal Array

Sample	А	В	С	D
<b>S</b> 1	1	1	1	1
S2	1	2	2	2
<b>S</b> 3	1	3	3	3
<b>S</b> 4	2	1	2	3
S5	2	2	3	1
<b>S</b> 6	2	3	1	2
S7	3	1	3	2
<b>S</b> 8	3	2	1	3
S9	3	3	2	1

Factors	Symbol	Levels		
		1	2	3
Guide Layers (µm)	А	1	2	3
Annealing Temp. (°C)	В	300	450	500
Annealing Time (hours)	С	2	4	6
Dye Loading Time (hours)	D	2	4	8

2.3 Preparation of Anode and Fabrication of Solar Cells.

Fluorine doped tin oxide glasses with sheet resistance 10  $\Omega$ /square and average transmittance of 85% were cut into 2  $cm \times 2$  cm size. The glasses were rinsed with alkali-free detergent solution for 5 minutes followed by deionized (DI) water. They were sonicated in steps using 0.1M HCl, acetone and isopropanol for 5 minutes each, and then finally boiled in isopropanol at 80°C for another 5 minutes. This was done to remove all residual contaminants from the surface of the glasses.Ethanol was added to titanium oxide powder in drops and stirred with plastic spatula until a slightly runny paste was obtained. The paste was deposited on the conductive sides of clean conductive glasses by doctor blading technique using aclean glass rod. They were preheated at 60°C for 5 minutes and then annealed at different temperature for different time. They were thereafter soaked in the prepared henna dye to obtain nine working electrodes as specified in Table 2 and Table 3. The film area was measured and the thickness was determined by mass difference technique before sensitization. The counter electrodes were made by

shading the conductive sides of another set of clean conductive glasses to deposit a layer of graphite as catalyst for current collection. Each of the dyed TiO<sub>2</sub> layers and each graphite coated counter electrode were assembled to form a solar cell by sandwiching a redox ( $I^-/I^3$ ) electrolyte solution. The electrolyte contains 0.025 g of Iodine and 0.025g of Potassium Iodide.

## 2.3. Characterization

The absorbance spectra of the dye samples in the wavelength range of 300 nm – 700 nm was studied with UV-VIS Spectrophotometer. The J-V characterization of the fabricated solar cell was carried out with Keithley 2400 series source meter under constant illumination of 1 Sun (100mW/cm<sup>2</sup>) at A.M. 1.5 from a solar simulator. The fill factorFF and overall conversion efficiency] were calculated using Eq. (1) and Eq. (2), respectively.

$$FF = \frac{l_m \times V_m}{l_{sc} \times V_{oc}} \tag{1}$$

$$\eta = J_{sc}(\text{mAcm}^{-2}) \times V_{oc}(V) \times FF$$
(2)

The shunt and series resistances were determined from I-V curves. The shunt resistance is the inverse of the slope at  $I_{sc}$  while the inverse of the slope at  $V_{oc}$  is proportional to the series resistance of a solar cell.

#### 2.4. Grey Relational Analysis (GRA)

In grey relational analysis (GRA), a suitable orthogonal array is selected, experimental results are analyzed, optimal combination is chosen, global parameters optimal performance isprojected and then experimentally projected improvement is verified. To do this, the signal-to-noise ratios of the responses are normalized between 0 and 1 based on the requirement [21]. optimization In this case, the responsesbecomedimensionless and consequently comparable in magnitude. For this study, the cell efficiency and shunt resistance are of the higher-the-better (HB) type while the series resistance is of the lower-the-better (LB) type. Since a higher value of efficiency is desirable, thus the higher-thebetter is chosen. Figure 1 is the equivalent circuit of solar cell which is considered as a current source (I<sub>L</sub>) in parallel with a forward biased diode  $(I_D)$ . Series  $(R_S)$  and parallel  $(R_{SH})$ resistances are added to account for various loss mechanisms [26]. Series resistance reduces the fill factor and short-circuit current (I), consequently the maximum power output is reduced. The open-circuit voltage (V) is not affected by R<sub>s</sub> since at V, the total current flow through the cell and

hence through the series resistance is zero. Low shunt resistance ( $R_{SH}$ ) provides an alternate current path for the photo-generated current leading to a significant power output loss. So, for higher efficiency, low  $R_S$  and high  $R_{SH}$  are desirable.



Figure 1: Equivalent circuit of solar cell [26]

The signal-to-noise ratios (SNR) of the HB and LB responses were determined from Eq. (3) and Eq. (4), respectively [22].

$$SNR_{HB} = -10log_{10}\left[\left(\frac{1}{n}\right)\left(\sum_{Y_{ij}^2}\right)\right]$$
 (3)

$$SNR_{LB} = -10 \log_{10} \left[ \sum_{n=1}^{\frac{Y_{ij}^2}{n}} \right]$$
 (4)

The normalization equations for the HB and LB responses are given by Eq. (5) and Eq.(6), respectively.

$$Z_{ij} = \frac{Y_{ij} - \min(Y_{ij}, i=1,2,..n)}{\max(Y_{ij}, i=1,2,..n) - \min(Y_{ij}, i=1,2,..n)}$$
(5)

$$Z_{ij} = \frac{\max(Y_{ij}, i=1, 2, .., n) - Y_{ij}}{\max(Y_{ij}, i=1, 2, .., n) - \min(Y_{ij}, i=1, 2, .., n)}$$
(6)

 $Z_{ij}$  is the normalized jth S/N ratio in the ith experiment and  $y_{ij}$  is the value of the response `j` in the ith experiment, max  $y_{ij}$  and min  $y_{ij}$  are the maximum and minimum values of signal-to-noise ratios for the jth response.

The grey relational coefficients were determined with Eq. (7):

$$\xi_i(k) = \frac{\Delta_{min}(k) + \zeta \Delta_{max}(k)}{\Delta_{0i}(k) + \zeta \Delta_{max}(k)}$$
(7)

 $\zeta_i(k)$  is the grey relational coefficient for the ith experiment for kth response.  $\Delta_{\partial i}(k)$  is the deviation sequence of the kth response of ith experiment. $\Delta_{\min}(k)$  and  $\Delta_{\max}(k)$  are the minimum and maximum deviation sequence of kth response. $\zeta$  is the distinguishing coefficient. Based on the importance of each response with reference to its contribution

to a good solar cell,  $\zeta$  was assigned values of 0.8, 0.15 and 0.05 for efficiency, series resistance, and shunt resistance, respectively. This indicates that the cell efficiency is more important than the series and shunt resistance for an efficient solar cell. Hence, series and shunt resistanceare given less attention during optimization.

The grey relational for the ith experiment grade  $\gamma_i$  was determined by Eq. (8):

$$\gamma_i = \frac{1}{n} \sum_{k=1}^n \xi_i(k) \tag{8}$$

#### 3. Results and Discussions

Results of the absorbance of the samples, I-V curves, GRA, optimization results and validation are presented in this section.

## 3.1. Absorbance Spectra of Dye Samples

From the absorbance spectra of the four dye samples (Figure 2), peak absorbance values of 2.882, 1.453 and 0.988 were obtained for raw fresh henna leave, industrial black henna and industrial red henna extracts, respectively at wavelength 440 nm. However, peak absorbance value of 2.069 was recorded at wavelength 540 nm for Ogbomoso black henna. The absorbance of the dye extracted from fresh henna leaf is higher than those of extracts from other industrially processed leaves at all wavelengths. This is attributed to the fact that some useful light absorbing components could have been lost or degraded as a result of the processing technique employed in industries and chemical additives to suit skin beautification.

Table 4: Thicknesses and J-V Reponses of DSSCs



Figure 2: Absorbance spectra of dye samples



Figure. 3: I-V Characteristic curves of DSSC

Sample	t (µm)	Voc (mV)	Jsc (mA/cm <sup>2</sup> )	FF	Ŋ(%)	Shunt R ( $\Omega$ )	Series R ( $\Omega$ )	
SS1	2.317	95	0.139	0.9034	0.0119	5090.9	16.24	
SS2	2.186	96	0.159	0.8296	0.0127	1323.8	10.96	
SS3	2.217	152	0.314	0.7627	0.0364	4436.4	6.43	
SS4	5.922	380	0.609	0.9000	0.2084	3771.0	13.04	
SS5	5.476	187	0.271	0.9017	0.0457	4014.3	7.36	
SS6	6.108	192	0.332	0.8594	0.0548	2391.8	23.06	
SS7	9.824	690	1.006	0.9437	0.6548	22100.0	9.44	
SS8	9.765	485	0.816	0.8542	0.3380	7297.7	24.98	
SS9	9.436	426	0.672	0.8922	0.2554	16800.0	10.45	
May	0.824	600	1.006	0.0427	0 65 19	22100.0	4205.00	
Min.	9.824 2.186	95	0.139	0.9437 0.7627	0.0348	1323.8	24.978	

Using Eq. (3)-(7), the grey relational analysis for efficiency, shunt resistance and series resistance were computed. The grey relational coefficients and grades for the nine experimental runs are presented in Table 5.Since experiment 7 has the highest grey relational grade (GRG) of 0.7234, it was selected as experimental best conditionwhich is coded as  $A_3B_1C_3D_2$  (Table 3). This condition corresponds to average thickness of 9.67µm, annealing temperature of 300°C, annealing time of 6 hours and dye loading time of 4 hours. The average GRGs for different levels of the fabrication parameters is presented in Table 6 and the equivalent effect plot in Figure 4. By selecting the maximum GRG for each parameter, the global best fabrication condition is A<sub>3</sub>B<sub>1</sub>C<sub>1</sub>D<sub>2</sub> with equivalent GRG of 0.7755. This corresponds to average thickness of 9.675 µm, temperature of 300°C, annealing time of 2 hours and dye loading time of 4 hours. This indicates that with optimization, the annealing time is reduced by 67%.

**Table 5**: Grey Relational Coefficients and Grades of

 Experiments

<b>a</b> 1	Grey R	Grey Relation		
Sample	η (%)	$R_{sh}\left(\Omega\right)$	$R_{s}\left(\Omega\right)$	al Grades
SS1	0.4444	0.0875	0.3209	0.2843
SS2	0.4481	0.0476	0.1982	0.2313
SS3	0.5257	0.0806	0.1304	0.2456
SS4	0.7367	0.0737	0.2384	0.3496
SS5	0.5461	0.0762	0.1428	0.2551
SS6	0.5637	0.0595	0.7180	0.4471
<u>SS7</u>	<u>1.0000</u>	<u>1.0000</u>	<u>0.1730</u>	<u>0.7243</u>
SS8	0.8289	0.1127	1.0000	0.6472
SS9	0.7729	0.3392	0.1893	0.4338

Parameters/Levels	(GRGs) <sub>ave</sub>		
	1	2	3
Thickness A	0.2537	0.3506	0.6018
Annealing Temp B	0.4527	0.3779	0.3755
Annealing Time C	0.4595	0.3382	0.4083
Dye Loading Time D	0.3244	0.4676	0.4141



**Figure 4**: Effects plot for selection of optimum parameter levels. An optimum level for a deposition parameter denotes the level at which the (GRG)<sub>ave</sub> is maximum.



**Figure 5**: I-V Characteristic Curves for experimental best and optimized DSSCs

#### 3.3 Validation of Optimization Procedure

After the optimal condition was applied to fabricate a DSSC, its I-V characteristic curve is compared to that of experimental best DSSC(Figure 5). Other responses are summarized in Table 7. The validated optimalDSSC has GRG of 0.866 compared to the projected value of 0.772. This translates to 1.3% higher efficiency, 12.4% higher shunt resistance and 10.8% lower series resistance. The average low value of the efficiency of the optimized DSSC can be attributed to the dye chosen for sensitization during the fabrication procedure. The effectiveness of henna dye as sensitizer can be improved by working on the dye structure for better recombination dynamics, co-sensitization with other dyes or dye additives to improve optical absorption throughout UV-Visible and near infra-red regions of the electromagnetic spectrum.

**Table 7:** Comparison between Experimental Best DSSC and

 Optimized DSSC

Response	Experimental	Best	Optimized
Response	57)		DSSC
$R_{sh}\left(\Omega ight)$	22100.00		24844.44
$R_s(\Omega)$	9.4433		8.4281
Voc	690		681
Jsc	1.006		1.036
FF	0.9437		0.9397
η%	0.6548		0.6630

## 4. Conclusion

The responses of TiO<sub>2</sub> based henna sensitized DSSCs were successfully optimized using grey-Taguchi technique. Based on the thickness of the working electrode, annealing temperature, annealing time and dye loading time, the optimal fabrication conditions were simultaneously determined for best combination of efficiency, shunt resistance and series resistance. The optimization procedure resulted in a DSSC with efficiency, shunt resistance and series resistance of 0.663%, 24844.44 $\Omega$  and 8.43 $\Omega$ , respectively. This indicates that the solar cell efficiency is increased by

1.3%, shunt resistance is enhanced by 12.4%, series resistance is lowered by 10.8% and annealing time is reduced by 67%.

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